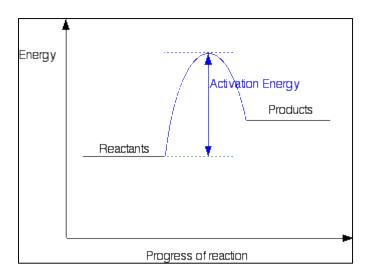
## MINISTRY OF EDUCATION, HERITAGE AND ARTS YEAR 12 CHEMISTRY REVISION WORKSHEET 7

Write the answers to the following questions in your exercise/activity books.

## **Strand 3: Reactions**

1. Study the diagram given below and answer the questions that follow.



Source: https://www.chemguide.co.uk

- (i). State whether the energy profile diagram shown above is of an endothermic or exothermic reaction. (1 mark)
- (ii). Explain your answer to part (i) above. (2 marks)
- 2. The reaction between zinc metal and hydrochloric acid solution gives out heat.

$$Zn_{(s)} \quad + \quad 2HCl_{(aq)} \quad \rightarrow \quad ZnCl_{2(aq)} \quad + \quad H_{2(g)}$$

- (i). State the sign of the  $\triangle_r H$  for the above reaction. (1 mark)
- (ii). Sketch an energy diagram for the reaction.

  Note: Label the reactants and products on the diagram. (4 marks)

## THE END